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determine the number of moles
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the following: a. 0.20 mol $\text{N}_2(\text{g})$ reacts
completely with $\text{H}_2(\text{g})$. b. 0.30 mol
 $\text{H}_2(\text{g})$ reacts completely with $\text{N}_2(\text{g})$. 2.
Considering the reaction represented
by the (unbalanced) equation

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stable isotopes of iron with mass
numbers 54, 56, 57, and 58 There are
two radioactive

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